**Concentration Gradient**

Watch the video multiple times and use the word bank to fill in the blank.

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| Concentration low gradient high down move stable down gradient  |

The concentration gradient went from \_\_\_\_\_\_\_\_\_\_ concentration on the left to \_\_\_\_\_\_\_\_\_\_\_concentration on the right. And so if you let some time pass, then they become more uniformly spread across a container. They have moved \_\_\_\_\_\_\_\_\_\_\_\_\_\_their concentration gradient to make things more uniform. The whole point of the video is to show that each particle moves down its unique concentration gradient. The yellow particles moves from high \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ to low \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_. They are going to \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ from right to left. And they’ll keep doing that until you get a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ configuration. So the real takeaway, you’ll hear in a biology or chemistry class, of things moving \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ their concentration \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.